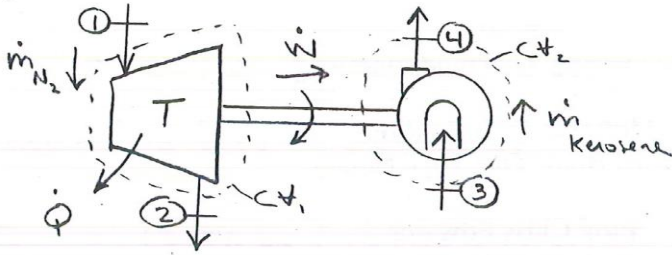


Aufgabe 1 (10 Punkte):

1. An exhaust turbine is used to drive a fuel pump.



$$T_1 = 1000\text{K}, T_2 = 400\text{K}$$

$$\dot{m}_{N_2} = 10\text{g/s}, \dot{Q} = 1\text{kW}$$

$$T_3 = T_4 = 20^\circ\text{C}, P_3 = 100\text{kPa}$$

$$P_4 = 1000\text{kPa}, \underline{\dot{W} = ?}, \underline{\dot{m}_{\text{keosene}} = ?}$$

$$\underline{\dot{W}} \quad [E]_{cv_1}: \frac{dE}{dt} = \dot{m}_1 h_1 - \dot{m}_2 h_2 - \dot{Q} - \dot{W}$$

Assume steady

$$\dot{W} = \dot{m}(h_1 - h_2) - \dot{Q} \quad \text{Assume Ideal Gas for } N_2$$

$$\dot{W} = \dot{m}_{N_2} c_{p,AVG} (T_1 - T_2) - \dot{Q} \quad \text{Choose } c_{p,N_2} \text{ at } T_{AVG} = 700\text{K}$$

$$\dot{W} = (10\text{g/s})(1.098 \frac{\text{kJ}}{\text{kg}\cdot\text{K}})(1000 - 400)\text{K} - 1\text{kW} \quad c_{p,AVG} = 1.098 \frac{\text{kJ}}{\text{kg}\cdot\text{K}}$$

$$\dot{W} = 5.59 \text{ kW}$$

$$\underline{\dot{m}_{\text{keosene}}} \quad [E]_{cv_2}: 0 = \dot{m}_{\text{keosene}} (h_3 - h_4) + \dot{W}$$

$$\dot{m}_{\text{keosene}} = \frac{\dot{W}}{h_4 - h_3}$$

$$\text{ICS: } \Delta h = \Delta u + v \Delta P$$

$$u = u(T) \text{ so } \Delta u = 0 (T_4 = T_3)!$$

$$\dot{m}_{\text{keosene}} = \frac{\dot{W}}{v(P_4 - P_3)}$$

$$v = \frac{1}{\rho_{\text{keosene}}} = \frac{1}{820 \text{ kg/m}^3} = 0.00122 \frac{\text{m}^3}{\text{kg}}$$

$$\dot{m}_{\text{keosene}} = \frac{5.588 \text{ kW}}{(0.00122 \text{ m}^3/\text{kg})(1000\text{kPa} - 100\text{kPa})} = 5.09 \text{ kg/s}$$

Alternative for Δh : $h_{\text{ICS}}(T, P) = h_f(T) + v_f(T)[P - P_{\text{SAT}}(T)]$

$$h_4 = h_f(T_4) + v_f(T_4)[P_4 - P_{\text{SAT}}(T_4)]$$

$$h_3 = h_f(T_3) + v_f(T_3)[P_3 - P_{\text{SAT}}(T_3)]$$

} But $T_3 = T_4 = 20^\circ\text{C}$

$$h_4 - h_3 = v_f(20^\circ\text{C})(P_4 - P_3)$$

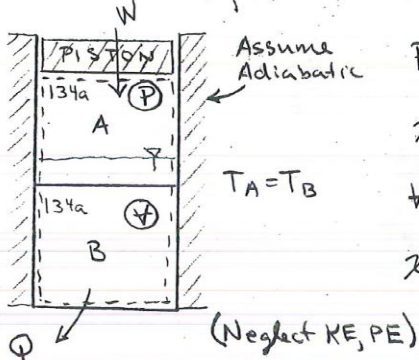
But $v_f = 1/\rho$ as above you get the same result.

$$\Delta h = v \Delta P!$$

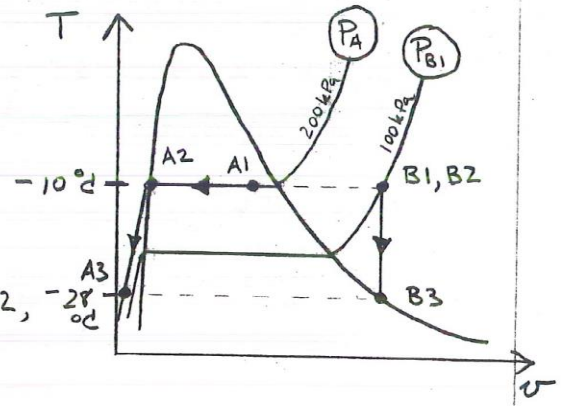
Aufgabe 2 (30 Punkte):

2. A piston-cylinder contains a rigid, diathermal wall. The piston floats freely while heat is extracted from the bottom in two steps. Find: $m_A, m_B, W_{12}, Q_{13}, T, U, T_3$.

2/3



Assume Adiabatic $P_A = 200 \text{ kPa}, x_{A1} = 0.8, v_{A1} = 1 \text{ m}^3$
 $x_{A2} = 0 \leftarrow (\text{Step } 1 \rightarrow 2)$
 $v_B = 1 \text{ m}^3, T_A = T_B, P_{B1} = 100 \text{ kPa}$
 $x_{B3} = 1 \leftarrow (\text{Step } 2 \rightarrow 3)$



Since $T_A = T_B$, the temperature in B cannot change until all of the vapor has disappeared in A.

Since T_B & v_B are fixed from $1 \rightarrow 2$, its state does not change.

Once A is outside the vapor dome, its temperature can change (at constant P), which allows the state in B to change.

m_A $v_{A1} = (1-x_{A1})v_f + x_{A1}v_g = (0.2) \cdot 0.0007532 \text{ m}^3/\text{kg} + (0.8) \cdot 0.0993 \text{ m}^3/\text{kg}$
 $v_{A1} = 0.07959 \text{ m}^3/\text{kg}, m_{A1} = v_{A1}/v_{A1} = 1 \text{ m}^3 / 0.07959 \text{ m}^3/\text{kg} = \boxed{12.56 \text{ kg}}$

m_B $T_{B1} = T_{A1} = -10^\circ\text{C}$ } Superheated
 $P_{B1} = 100 \text{ kPa}$ } $v_{B1} = 0.20686 \text{ m}^3/\text{kg}$ $m_{B1} = \frac{v_B}{v_{B1}} = \frac{1 \text{ m}^3}{0.20686 \text{ m}^3/\text{kg}} = \boxed{4.834 \text{ kg}}$

W_{12} Constant P! $W_{12} = -P(v_2 - v_1) = -P m_A (v_{A2} - v_{A1})$
 $v_{A2} = v_f(T_{A2}) = 0.0007532 \text{ m}^3/\text{kg}, W_{12} = (200 \text{ kPa})(12.56 \text{ kg}) \times (0.0007532 - 0.07959) \text{ m}^3/\text{kg}$
 $\Rightarrow \boxed{W_{12} = 198.0 \text{ kJ}}$ \leftarrow Work done on the refrigerant!

T_3 $T_3 = T_{\text{SAT}}$ at same specific volume as B1. $v_{B3} = 0.20686 \text{ m}^3/\text{kg} = v_g \approx 0.2052$ (1%)

Q_{13} [E]: $\Delta U = W_{13} - Q_{13} \Rightarrow Q_{13} = -\Delta U_A - \Delta U_B + W_{12}$ $\left\{ \begin{array}{l} \text{Assume } \Rightarrow T_3 = -28^\circ\text{C} \\ W_{23} = 0 \\ \text{since it is impossible!} \end{array} \right.$
 $Q_{13} = m_A(u_{A1} - u_{A3}) + m_B(u_{B1} - u_{B3}) + W_{12}$

$$u_{A1} = (1-x_{A1}) u_f + x_{A1} u_g = (0.2) 36.69 \frac{\text{kJ}}{\text{kg}} + (0.8) 221.43 \frac{\text{kJ}}{\text{kg}} = 184.48 \frac{\text{kJ}}{\text{kg}}$$

$$u_{A3} = u_f(T_{A3}) \text{ (ICS!)} \Rightarrow u_{A3} = u_f(-28^\circ\text{C}) = 14.31 \frac{\text{kJ}}{\text{kg}}$$

$$u_{B1} = 224.01 \frac{\text{kJ}}{\text{kg}} \text{ (Superheated)}$$

$$u_{B3} = u_g(T_3) = u_g(-28^\circ\text{C}) = 211.29 \frac{\text{kJ}}{\text{kg}}$$

$$Q_{13} = 12.56 \text{ kg} (184.48 - 14.31) \frac{\text{kJ}}{\text{kg}} \quad \{ 2137 \text{ kJ}$$

$$+ 4.834 \text{ kg} (224.01 - 211.29) \frac{\text{kJ}}{\text{kg}} \quad \{ 61.5 \text{ kJ}$$

$$+ 198.0 \text{ kJ} \quad \{ 198 \text{ kJ}$$

$$Q_{13} = 2396.5 \text{ kJ}$$

Note: If you wished to include the work from 2→3, it would

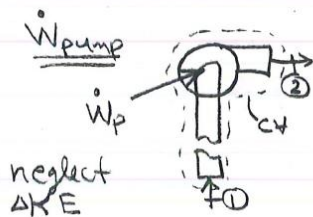
$$\text{amount to } W_{23} = m_A P_A (v_f(-10^\circ\text{C}) - v_f(-28^\circ\text{C}))$$

$$= (12.56 \text{ kg}) (200 \text{ kPa}) (.0007532 - .0007233) \frac{\text{m}^3}{\text{kg}}$$

$$W_{23} = 0.075 \text{ kJ} \quad \text{Not a significant contribution} \\ (.003\%!)$$

Aufgabe 3 (10 Punkte):

3. Brayton & Pump



$$[E]: 0 = \dot{W}_p + \dot{m}(h_1 + gz_1) - \dot{m}(h_2 + gz_2)$$

$$\dot{W}_p = \dot{m}[(h_2 - h_1) + g(z_2 - z_1)]$$

$$\text{ICS} \Rightarrow \Delta h = \Delta u + v \Delta P = 0$$

(Rev) 0 same pressure

$$\Rightarrow \dot{W}_p = \dot{m} g(z_2 - z_1) = (10 \frac{\text{kg}}{\text{s}}) (300 \text{ m}) (9.8 \frac{\text{m}}{\text{s}^2})$$

$$\dot{W}_p = 29.4 \text{ kW}$$

$$\text{Brayton} \Rightarrow \eta_{th} = 1 - \left(\frac{1}{r_p}\right)^{\frac{k-1}{k}}, \quad k=1.4, \quad r_p=9 \Rightarrow \eta_{th} = 46.6\%$$

$$\eta_{th} = \frac{\dot{W}}{\dot{Q}} = \frac{\dot{W}_p}{\dot{Q}} \Rightarrow \dot{Q} = \frac{\dot{W}_p}{\eta_{th}} = \frac{29.4 \text{ kW}}{.466} = 63.1 \text{ kW}$$

Aufgabe 4 (40 Punkte):

a) siehe Diagramm

b) Da der Wasserabscheider zwar mit Druckverlust aber adiabat arbeitet, bleibt die spezifische Enthalpie erhalten:

⇒ isenthalpe Zustandsänderung

c) Wasserabscheider:

$$\text{Dampfgehalt: } x_2 = \frac{h_2 - h_4}{h_3 - h_4}, \quad h_2 = h_4$$

aus Tab.:

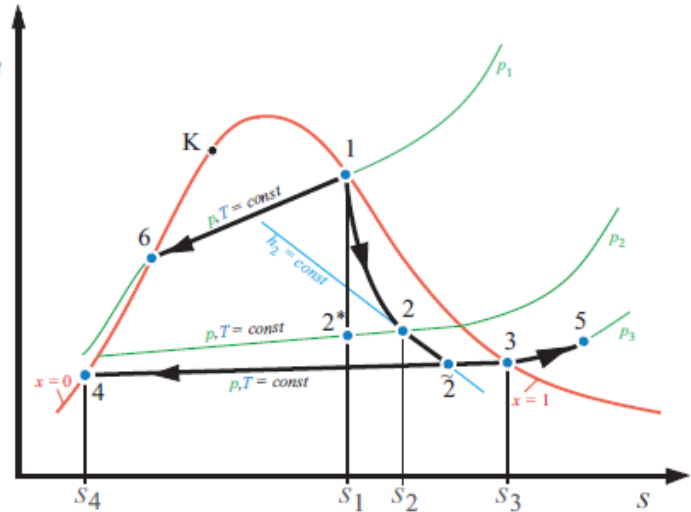
$$h_4 = h(p_3, x_3) = h'(p_3)$$

$$h_3 = h(p_3, x_3) = h''(p_3)$$

Für die Turbine gilt:

$$\eta_{s,T} = \frac{h_2 - h_1}{h_{2^*} - h_1} \Rightarrow h_2 (= h_1 + \eta_{s,T} (h_{2^*} - h_1))$$

$$\text{aus Tab.: } h_{2^*} = h(p_2, s_{2^*}), \quad s_{2^*} = s_1 = s(p_1, x_1) = s''(p_1), \quad h_1 = h(p_1, x_1) = h''(p_1)$$



d) Massenbilanz Wasserabscheider: $\frac{\dot{m}_{III}}{\dot{m}_{IV}} = \frac{x_2}{1 - x_2}$

e) Energiebilanz Überhitzer (stationärer Fließprozess):

$$\left. \begin{aligned} \dot{m}_I h_1 + \dot{m}_{III} h_3 &= \dot{m}_5 h_5 + \dot{m}_6 h_6, \quad \dot{m}_5 = \dot{m}_{III}, \quad \dot{m}_6 = \dot{m}_I \\ \text{aus Tab.: } h_1, h_3 &= \text{s.o.}, \quad h_5 = h(p_3, \vartheta_5), \quad h_6 = h(p_1, x_6) = h'(p_1) \end{aligned} \right\} \Rightarrow \frac{\dot{m}_I}{\dot{m}_{III}} \left(= \frac{h_5 - h_3}{h_1 - h_6} > 0 \checkmark \right)$$

f) Massenbilanz Turbine:

$$\dot{m} = \dot{m}_I + \dot{m}_{II} = \dot{m}_{III} \frac{h_5 - h_3}{h_1 - h_6} + \dot{m}_{II} = \dot{m}_{II} \left(x_2 \frac{h_5 - h_3}{h_1 - h_6} + 1 \right) \Rightarrow \dot{m}_{II} \left(= \frac{\dot{m}}{1 + x_2 \frac{h_5 - h_3}{h_1 - h_6}} \right)$$

g) Energiebilanz Turbine:

$$\dot{Q}_{12} + P_{12} = \dot{H}_2 - \dot{H}_1 = \dot{m}_{II} (h_2 - h_1), \quad Q_{12} = 0 \Rightarrow P_T = \dot{m}_{II} (h_2 - h_1) (< 0 \checkmark)$$

h) Entropiebilanz Wasserabscheider (stationärer Fließprozess):

$$\frac{dS}{dt} = 0 = \dot{S}_2 - \dot{S}_3 - \dot{S}_4 + \dot{S}_Q + \dot{S}_{irr}, \quad \dot{S}_Q = 0 \Rightarrow \dot{S}_{irr} = \dot{m}_{III} s_3 + \dot{m}_{IV} s_4 - \dot{m}_{II} s_2$$

$$\text{aus Tab.: } s_3 = s(p_3, x_3) = s''(p_3), \quad s_4 = s(p_3, x_4) = s'(p_3), \quad s_2 = s(p_2, h_2)$$